

## Research in the Nienhaus Lab



Lea Nienhaus Florida State University Department of Chemistry and Biochemistry

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### Introducing the Nienhaus Lab (Est. 2018)





### **Photon Upconversion**

Upconversion: combining two or more low energy photons to one higher energy photon.



In our system upconversion occurs through triplettriplet annihilation in organic semiconductors.





# **Triplet-Triplet Annihilation**

Taking advantage of non-emissive, long-lived triplet states in polyacenes. Two anti-correlated triplets interact without forbidden spin flip.







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How do we efficiently access the triplet state?



### **Triplet Sensitizers**



#### Perovskites





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# Challenges: Quantum-Confined Triplet Sensitizers

d Areas of Improvement:



Maximize apparent anti-Stokes shift

느냐???

Pathways



### Nanoplatelet Sensitizers







### Nanoplatelet Sensitizers

Increasing the dimensionality:  $0D \rightarrow 2D$ 



Despite the promising lower  $I_{th}$ , we see a lower UC QY for NPL sensitized UC.





### Nanoplatelet Sensitizers





- Stacking can cause inaccessible transmitter ligands → not TET active.
- Stacking can reduce the sites transmitter ligands can bind to.



### Lateral Dimensions – Size Matters!

 NPLs are only confined in one dimension
→ change size/absorptivity without changing energetics







### Lateral Dimensions – Size Matters!





Reduction in the UC QY with increasing size is not simply due to lower PLQY:

2-fold reduction in the PLQY but 5-fold reduction in the UC QY





### Outlook



MA

NPLs are promising sensitizers in solid state due to the promise of rapid and efficient long-range energy transfer.

Requirement for solid-state UC device fabrication: efficient solid-state annihilators

Calla Crara 116



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### **Perovskite-Sensitized Triplet Generation**







## Near-Infrared-to-Visible Upconversion

Converting infrared (low energy) light to visible (high energy) light via sensitized triplet-triplet annihilation.



Photo: Dan Congreve

### 800 nm 610 nm 610 nm 610 nm h<sup>+</sup> h<sup>+</sup> h<sup>+</sup> h<sup>+</sup>

Nienhaus, L., et al. ACS Energy Lett. 2019

**Perovskite-based upconversion** Current benchmark: up to 60% absorption

QD-based upconversion

Limited by poor exciton diffusion Current benchmark: <1% absorption



Perovskites are very susceptible to their environment. Solvent treatment can remove individual ions, resulting in a chance in the local surface composition, which will change the electronic structure.

Solvents classified according to Taylor et al.: (*Nat. Commun.* 12, 1878 (2021))

Type I: dissolve MAI/FAI  $\rightarrow$  PbI<sub>2</sub> generation Type II: do not dissolve MAI/FAI or PbI<sub>2</sub> Type III: generate I<sub>3</sub><sup>-</sup>

ACN: dissolves both MAI/FAI and Pbl<sub>2</sub> once sufficient MAI/FAI is dissolved.





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In agreement with the increased UC PL intensity, we find higher amounts of quenching for Type I solvents, while Type II and Type III solvents reduce the amount of quenching.





XRD shows removal of the undesired  $\delta$ -phase with Type I solvents and ACN.

Does the removal of individual ions result in a change in the surface morphology? Is a 2D perovskite structure grown over the 3D perovskite? Is there a rearrangement at the surface?





The presence of defects and PbI<sub>2</sub> influence the electronic structure of the perovskite surface.

Defects and  $PbI_2$  can induce doping, hence affect the interfacial band bending.

The change in  $E_F$  will greatly influence the band alignment at the perovskite/rubrene interface and thus, the charge transfer.

We propose that interfacial defects do not inherently mediate the charge transfer but result in improved charge extraction due to the interfacial energy alignment.









The presence of defects and Pbl<sub>2</sub> influence the electronic structure of the perovskite surface.

n-type doping results in hole accumulation at the surface.

p-type doping results in holes accumulating in the bulk.

 $\rightarrow$  Changes in the interfacial charge extraction





### **Next-Generation Annihilators**





### **Next-Generation Annihilators**



Due to the limited achievable apparent anti-Stokes shift (800 nm  $\rightarrow$  565 nm/605 nm) the rubrene/DBP pair is not well matched to the iodide-based perovskite.

Ideally, an annihilator with  $T_1 \approx 1.5$  eV is desired.

However, many successful solution-based annihilators do not show the same results in solid state (e.g., DPA is plagued by excimer formation).

1-chloro-9,10-bis(phenylethynyl)anthracene ( $T_1$  = 1.2 eV is a promising step towards expanding the compatible solid-state annihilator library.



### **Next-Generation Annihilators**

PL quenching is promising, but not sufficient to determine whether the triplet is created.

PL quenching could also indicate single charge (hole) extraction.







## Summary and Outlook

Progress is being made in understanding triplet generation at the perovskite/organic interface.

The mechanism of triplet generation based on free carrier injection has been understood.

New annihilators have been identified. More to come, stay tuned...

Open questions: role of nanoscale OSC arrangement and coupling. Aggregationinduced effects? Strong coupling vs. weak coupling?





# Kitchen Spectroscopy



from lab to kitchen



Glow from home: C&EN Chem Pics May 2020



### FLORIDA STATE UNIVERSITY



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@NienhausFSU